

Name of the Student: _____

Max. Marks : 17 Marks

Time : 17 Minutes

Q1.

The table gives the properties of some radionuclides (radioactive isotopes).

Radionuclide	Half life	Main type of radiation emitted
Radon-220	54.5 seconds	Alpha
Americium-241	433 years	Alpha
Phosphorus-32	14 days	Beta
Strontium-90	28 years	Beta
Technetium-99	6 hours	Gamma
Cobalt-60	5 years	Gamma

- (i) Which radionuclide would be best for monitoring the thickness of aluminium foil?

Explain the reason for your answer.

(2)

- (ii) Which radionuclide would be best for acting as a tracer inside the human body?

Explain the reason for your answer.

(2)**(Total 4 marks)**

Q2.

$^{99}_{43}\text{Tc}$ (technetium) is produced by the radioactive decay of $^{99}_{42}\text{Mo}$ (molybdenum).

What change occurs in the nucleus of a molybdenum atom when this happens?

(Total 1 mark)

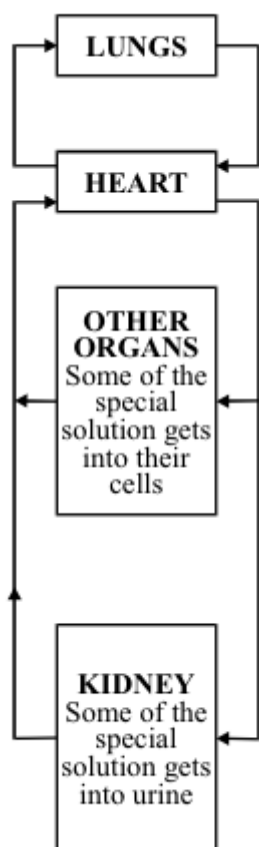
Q3.

Doctors sometimes need to know how much blood a patient has.

They can find out by using a radioactive solution.

After measuring how radioactive a small syringe-full of the solution is they inject it into the patient's blood.

YOUR BLOOD CIRCULATION



They then wait for 30 minutes so that the solution has time to become completely mixed into the blood.

Finally, they take a syringe-full of blood and measure how radioactive it is.

Example:

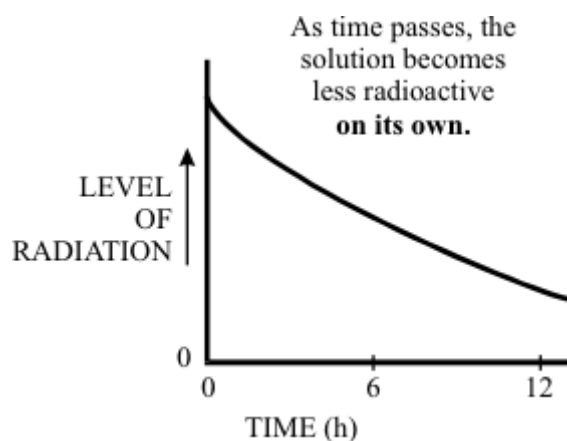
If the doctor injects 10 cm^3 of the radioactive solution and this is diluted 500 times by the blood there must be $10 \times 500 = 5000 \text{ cm}^3$ of blood.

(a) After allowing for background radiation:

- 10 cm³ of the radioactive solution gives a reading of 7350 counts per minute;
- a 10 cm³ sample of blood gives a reading of 15 counts per minute.

Calculate the volume of the patient's blood.
(Show your working.)

(4)



Radiation from radioactive substances can harm your body cells.

- (b) The doctor's method of estimating blood volume will not be completely accurate. Write down **three** reasons for this.

1. _____
2. _____
3. _____

(3)

- (c) The doctors use a radioactive substance which loses half of its radioactivity every six hours. Explain why this is a suitable radioactive substance to use.

(2)

(Total 9 marks)

Q3.

When atoms of uranium 238 (U^{238}) decay they produce another radionuclide called thorium 234 (Th^{234})

Thorium 234 (Th^{234}) decays by emitting beta radiation.

- (i) What does beta radiation consist of?

(1)

- (ii) Thorium 234 (Th^{238}) decays to form protactinium 234 (Pa^{234}).

What differences are there between the nucleus of a protactinium 234 (Pa^{234}) atom and the nucleus of a thorium 234 (Th^{234}) atom?

(2)

(Total 3 marks)