

Name of the Student: _____

Max. Marks : 17 Marks

Time : 17 Minutes

Mark Schemes

Q1.

(a) (i) B

for one mark

1

(ii) has 4 electrons / protons others only 3; B has a different no. of electrons / protons - not A and C have same no. of protons / electrons
for one mark

1



(b) (i) A and C

for one mark

1


(ii) same no. of protons / electrons different no. of neutrons
or
nuclei have the same charge but different mass
for 1 mark each

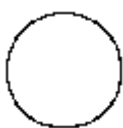

2

(c) (i) (ii) (iii) 

for 1 mark each

3

(d) $2p.2n$ 

allow  but not 
(i.e. no mark if electrons shown)
for one mark

1

[9]**Q2.**

(a) radium

accept Ra

1

- (b) different numbers of protons
 accept one has 91 protons, one has 92
or Pa has 91 protons, U has 92
 do **not** credit they have different atomic numbers
 reject different numbers of protons and neutrons

1

- (c) alpha

1

- (d) neutron changes into proton
 accept electron lost / beta radiation
 accept singular or plural answers

1

[4]

Q3.

- (a) (i) two protons

1

2 neutrons

if neither point gained allow 1 mark for helium nucleus

1

- (ii) electron

1

- (b) neutron splits (to form proton and electron)

1

[4]