

**Name of the Student:** \_\_\_\_\_

**Max. Marks : 18 Marks**

**Time : 18 Minutes**

Mark Schemes

**Q1.**

(a) 10 000 1

(b) **Increase**  
absorb electromagnetic radiation 1

**Decrease**  
emit electromagnetic radiation 1

(c) atomic number is the number of protons 1  
  
mass number is the number of protons and neutrons 1

(d) **Level 2 (3–4 marks):**  
A clear comparison, with logical structure.

**Level 1 (1–2 marks):**  
Fragmented points, with no logical structure.

**0 marks:**  
No relevant content

**Indicative content**

**Beta decay**

- Atomic number increases by one
- When a neutron decays into a proton

**Alpha decay**

- Atomic number decreases by two
- When an alpha particle is emitted

**Comparison**

Both change number of protons (hence new element / transmutation)  
Beta decay increases atomic number and alpha decay decreases (explicit)

NB No credit is given for different number of protons = new element.

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**Q2.**

- (a) (i) splitting of a(n atomic) nucleus  
*do not accept splitting an atom* 1
- (ii) Neutron 1
- (b) (i) nuclei have the same charge  
**or**  
nuclei are positive  
*accept protons have the same charge* 1
- (ii) (main sequence) star  
*accept Sun or any correctly named star*  
*accept red (super) giant* 1
- (c) (i) any **two** from:  
• easy to obtain / extract  
• available in (very) large amounts  
• releases more energy (per kg)  
*do **not** accept figures only*  
• produces little / no radioactive waste.  
*naturally occurring is insufficient*  
*seawater is renewable is insufficient*  
*less cost is insufficient* 2
- (ii) any **one** from:  
• makes another source of energy available  
• increases supply of electricity  
• able to meet global demand  
• less environmental damage  
• reduces amount of other fuels used.  
*accept any sensible suggestion*  
*accept a specific example*  
*accept a specific example* 1
- (d) 12  
*allow 1 mark for obtaining 3 half-lives* 2

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